Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

2-{4-[(Quinolin-8-yloxy)methyl]phenyl}-benzonitrile

Bin Wei

Ordered Matter Science Research Center, Southeast University, Nanjing 210096, People's Republic of China
Correspondence e-mail: seuwei@126.com

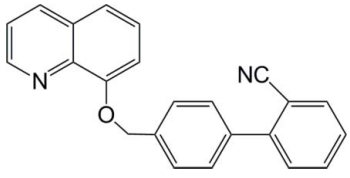
Received 31 March 2011; accepted 6 April 2011

Key indicators: single-crystal X-ray study; T = 293 K; mean σ(C–C) = 0.005 Å; R factor = 0.085; wR factor = 0.234; data-to-parameter ratio = 17.2.

In the title compound, $C_{23}H_{16}N_2O$, the bond angle at the O atom that connects the benzene ring and the quinoline ring system is 116.0 (2)°. The quinoline ring system make a dihedral angle of 16.5 (2)° with the adjacent benzene ring. The dihedral angle between the biphenyl benzene rings is 70.8 (2)°.

Related literature

For background to tetrazoles, see: Hang *et al.* (2009). For our investigation of tetrazole compounds and their coordination modes, see: Xiong *et al.* (2002). For the preparation of tetrazoles using *in situ* synthesis of tetrazole through cycloaddition between organotin azide and organic cyano groups, see: Chen *et al.* (2010); Ye *et al.* (2006).



Experimental

Crystal data

$C_{23}H_{16}N_2O$
M_r = 336.38
Orthorhombic, *Pbca*
a = 14.526 (4) Å
b = 8.957 (3) Å
c = 27.126 (8) Å
V = 3529.3 (18) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.08 mm^{-1}
T = 293 K
0.20 × 0.20 × 0.20 mm

Data collection

Rigaku Mercury CCD diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)
T_{min} = 0.842, T_{max} = 1.000
36455 measured reflections
4036 independent reflections
2865 reflections with $I > 2\sigma(I)$
R_{int} = 0.073

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.082
$wR(F^2)$ = 0.222
S = 1.24
4036 reflections
235 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.19 e Å^{-3}
$\Delta\rho_{min}$ = −0.22 e Å^{-3}

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

The author is grateful to the starter fund of Southeast University for the purchase of the diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: JH2278).

References

Chen, L. Z., Huang, Y., Xiong, R. G. & Hu, H. W. (2010). *J. Mol. Struct.* **963**, 16–21.
Hang, T., Fu, D. W., Ye, Q. & Xiong, R. G. (2009). *Cryst. Growth Des.* **5**, 2026–2029.
Rigaku (2005). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Xiong, R. G., Xue, X., Zhao, H., You, X. Z., Abrahams, B. F. & Xue, Z. L. (2002). *Angew. Chem. Int. Ed.* **41**, 3800–3803.
Ye, Q., Song, Y. M., Wang, G. X., Chen, K., Fu, D. W., Chan, P. W. H., Zhu, J. S., Huang, S. D. & Xiong, R. G. (2006). *J. Am. Chem. Soc.* **128**, 6554–6556.

supplementary materials

Acta Cryst. (2011). E**67**, o1209 [**doi:10.1107/S1600536811012724**]

2-{4-[(Quinolin-8-yloxy)methyl]phenyl}benzonitrile

B. Wei

Comment

Tetrazole compounds have been studied for more than one hundred years and applied in various areas (Hang *et al.*, 2009). As a part of systematic investigation of new tetrazole compounds and discovery of new coordination mode (Xiong *et al.*, 2002), we get the synthesis of the title compound C22 H16 N2 O,and preparation of tetrazoles *in situ* synthesis of tetrazole through cycloaddition between organotin azide and organic cyano group (Ye *et al.*, 2006; Chen *et al.*, 2010).

In the asymmetric unit of the title compound, the planes angle between the two benzene rings is 70.8°. O1 connect quinoline ring and sartan ring with a 115.9 bond-angle and the bond length O1—C10 is 1.4261 (35) Å, O1—C9 is 1.3691 (33) Å). The quinoline ring make a small dihedral angle of 16.5° with adjacent benzene ring (Fig 1). Fig 2 shows that the molecules assemble as straight chain in the crystal structure along the *a* axis.

Experimental

8-hydroxyquinoline(1.45 g,10 mmol) was added in a solution of 4'-Bromoethyl-2-cyanobiphenyl(2.71 g,10 mmol) in methanol(20 ml).After the mixture was stirred for 10 h at 355 K,the precipitate was filtered off and the solution was evaporated in vacuum. The crude product was then crystallized form ethanol to yield colourless prisms of the title compound.

Refinement

H atoms were positioned geometrically, with C—H = 0.93 and 0.97 Å for aromatic and methylene H respectively, and constrained to ride on their parent atoms with Uĩso~(H) = xU~eq~(C), where *x* = 1.5 for methyl H and *x* = 1.2 for all other H atoms.

Figures

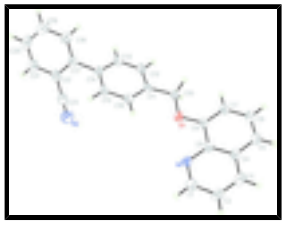


Fig. 1. Crystal structure of the title compound with labelling and displacement ellipsoids drawn at the 30% probability level.

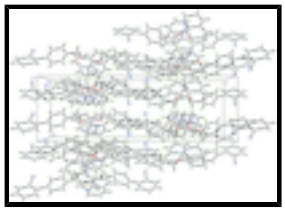


Fig. 2. Crystal structure of the title compound with view along the *a* axis. Intermolecular interactions are shown as dashed lines.

2-{4-[(Quinolin-8-yloxy)methyl]phenyl}benzonitrile

Crystal data

$C_{23}H_{16}N_2O$	$D_x = 1.266$ Mg m^{-3}
$M_r = 336.38$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Orthorhombic, *Pbca*	Cell parameters from 6593 reflections
$a = 14.526$ (4) Å	$\theta = 2.3$–$27.5°$
$b = 8.957$ (3) Å	$\mu = 0.08$ mm^{-1}
$c = 27.126$ (8) Å	$T = 293$ K
$V = 3529.3$ (18) Å^3	Prism, colorless
$Z = 8$	0.20 × 0.20 × 0.20 mm
$F(000) = 1408$	

Data collection

Rigaku Mercury CCD diffractometer	4036 independent reflections
Radiation source: fine-focus sealed tube	2865 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.073$
Detector resolution: 28.5714 pixels mm^{-1}	$\theta_{max} = 27.5°$, $\theta_{min} = 1.5°$
ω scans	$h = -18\rightarrow18$
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)	$k = -11\rightarrow11$
$T_{min} = 0.842$, $T_{max} = 1.000$	$l = -35\rightarrow35$
36455 measured reflections	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.082$	H-atom parameters constrained
$wR(F^2) = 0.222$	$w = 1/[\sigma^2(F_o^2) + (0.0817P)^2 + 0.8691P]$ where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.24$	$(\Delta/\sigma)_{max} < 0.001$
4036 reflections	$\Delta\rho_{max} = 0.19$ e Å^{-3}
235 parameters	$\Delta\rho_{min} = -0.22$ e Å^{-3}
0 restraints	Extinction correction: *SHELXL*
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.0000

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds

in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > 2\text{sigma}(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($Å^2$)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	0.28089 (13)	0.1887 (2)	0.29588 (6)	0.0503 (5)
N1	0.43511 (16)	0.3299 (3)	0.26974 (8)	0.0494 (6)
C8	0.38445 (18)	0.2633 (3)	0.23340 (9)	0.0421 (6)
C11	0.17557 (18)	0.1423 (3)	0.36270 (10)	0.0430 (6)
C10	0.2001 (2)	0.1096 (3)	0.30999 (10)	0.0501 (7)
H10A	0.2103	0.0032	0.3060	0.060*
H10B	0.1494	0.1383	0.2887	0.060*
C9	0.30312 (19)	0.1836 (3)	0.24685 (9)	0.0435 (6)
C4	0.4091 (2)	0.2693 (3)	0.18282 (10)	0.0481 (7)
C16	0.2179 (2)	0.2507 (4)	0.39095 (10)	0.0534 (7)
H16A	0.2675	0.3036	0.3781	0.064*
C17	0.0766 (2)	0.2387 (3)	0.50852 (10)	0.0502 (7)
C14	0.11330 (18)	0.2051 (3)	0.45827 (10)	0.0462 (6)
C7	0.2541 (2)	0.1086 (3)	0.21157 (11)	0.0542 (7)
H7A	0.2025	0.0537	0.2207	0.065*
C5	0.3561 (2)	0.1925 (4)	0.14748 (11)	0.0608 (8)
H5A	0.3727	0.1960	0.1144	0.073*
C22	0.1263 (2)	0.2008 (3)	0.55117 (10)	0.0540 (7)
C15	0.1872 (2)	0.2815 (4)	0.43839 (10)	0.0566 (8)
H15A	0.2167	0.3544	0.4570	0.068*
C13	0.0722 (2)	0.0939 (3)	0.42995 (11)	0.0536 (7)
H13A	0.0231	0.0399	0.4428	0.064*
C12	0.1033 (2)	0.0629 (3)	0.38308 (10)	0.0534 (7)
H12A	0.0753	−0.0125	0.3649	0.064*
C6	0.2808 (2)	0.1134 (4)	0.16163 (11)	0.0627 (9)
H6A	0.2465	0.0620	0.1382	0.075*
C2	0.5370 (2)	0.4211 (4)	0.20652 (13)	0.0682 (9)
H2A	0.5888	0.4776	0.1988	0.082*
C3	0.4879 (2)	0.3538 (4)	0.17066 (11)	0.0592 (8)
H3A	0.5059	0.3630	0.1379	0.071*
C1	0.5086 (2)	0.4045 (4)	0.25572 (12)	0.0626 (8)
H1A	0.5442	0.4492	0.2801	0.075*
C23	0.2152 (3)	0.1308 (4)	0.54694 (12)	0.0694 (9)
C21	0.0894 (3)	0.2274 (4)	0.59785 (12)	0.0677 (9)
H21A	0.1226	0.2013	0.6259	0.081*
C18	−0.0090 (2)	0.3026 (4)	0.51433 (13)	0.0685 (9)
H18A	−0.0434	0.3274	0.4866	0.082*
C20	0.0046 (3)	0.2917 (4)	0.60239 (14)	0.0755 (11)

H20A	−0.0200	0.3093	0.6335	0.091*
N2	0.2850 (3)	0.0735 (5)	0.54410 (13)	0.0980 (12)
C19	−0.0448 (3)	0.3306 (4)	0.56077 (16)	0.0808 (11)
H19A	−0.1021	0.3756	0.5639	0.097*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0526 (11)	0.0582 (12)	0.0403 (10)	−0.0130 (9)	0.0084 (8)	−0.0045 (8)
N1	0.0530 (14)	0.0500 (13)	0.0452 (13)	−0.0064 (11)	0.0065 (11)	−0.0081 (10)
C8	0.0484 (15)	0.0366 (13)	0.0413 (14)	0.0055 (11)	0.0039 (12)	−0.0010 (10)
C11	0.0430 (14)	0.0442 (14)	0.0419 (13)	−0.0008 (12)	0.0014 (11)	0.0029 (11)
C10	0.0508 (16)	0.0548 (17)	0.0446 (15)	−0.0085 (13)	0.0050 (12)	−0.0010 (12)
C9	0.0477 (15)	0.0438 (14)	0.0390 (13)	0.0023 (12)	0.0022 (11)	−0.0024 (11)
C4	0.0576 (17)	0.0451 (15)	0.0417 (14)	0.0078 (13)	0.0067 (13)	−0.0003 (11)
C16	0.0488 (16)	0.0679 (18)	0.0434 (15)	−0.0165 (14)	0.0076 (12)	0.0006 (13)
C17	0.0560 (17)	0.0472 (15)	0.0475 (16)	−0.0096 (13)	0.0073 (13)	0.0010 (12)
C14	0.0454 (15)	0.0493 (15)	0.0439 (14)	0.0000 (12)	0.0026 (12)	0.0031 (12)
C7	0.0549 (16)	0.0600 (18)	0.0477 (15)	−0.0070 (14)	0.0022 (14)	−0.0087 (13)
C5	0.071 (2)	0.073 (2)	0.0380 (15)	0.0056 (17)	0.0026 (14)	−0.0006 (14)
C22	0.0641 (19)	0.0542 (16)	0.0438 (15)	−0.0110 (15)	0.0077 (14)	−0.0029 (13)
C15	0.0585 (18)	0.0662 (19)	0.0452 (16)	−0.0195 (15)	0.0037 (13)	−0.0071 (14)
C13	0.0552 (17)	0.0577 (17)	0.0479 (15)	−0.0166 (14)	0.0076 (13)	0.0020 (13)
C12	0.0611 (18)	0.0536 (16)	0.0455 (15)	−0.0181 (14)	0.0011 (14)	−0.0025 (13)
C6	0.066 (2)	0.077 (2)	0.0458 (16)	−0.0022 (17)	−0.0065 (15)	−0.0132 (15)
C2	0.070 (2)	0.0615 (19)	0.073 (2)	−0.0180 (17)	0.0265 (18)	−0.0058 (17)
C3	0.072 (2)	0.0546 (17)	0.0515 (17)	−0.0007 (16)	0.0188 (16)	0.0033 (14)
C1	0.0637 (19)	0.0588 (18)	0.065 (2)	−0.0146 (16)	0.0060 (16)	−0.0129 (15)
C23	0.075 (2)	0.089 (3)	0.0438 (17)	0.001 (2)	−0.0009 (17)	0.0068 (16)
C21	0.089 (3)	0.066 (2)	0.0479 (17)	−0.0169 (19)	0.0109 (17)	−0.0081 (15)
C18	0.065 (2)	0.074 (2)	0.066 (2)	0.0056 (18)	0.0153 (17)	0.0038 (17)
C20	0.103 (3)	0.060 (2)	0.063 (2)	−0.014 (2)	0.036 (2)	−0.0132 (17)
N2	0.088 (3)	0.135 (3)	0.071 (2)	0.025 (2)	−0.0001 (19)	0.015 (2)
C19	0.080 (3)	0.070 (2)	0.092 (3)	0.005 (2)	0.036 (2)	−0.002 (2)

Geometric parameters (Å, °)

O1—C9	1.369 (3)	C5—C6	1.359 (5)
O1—C10	1.423 (3)	C5—H5A	0.9300
N1—C1	1.316 (4)	C22—C21	1.395 (4)
N1—C8	1.367 (3)	C22—C23	1.441 (5)
C8—C4	1.419 (4)	C15—H15A	0.9300
C8—C9	1.427 (4)	C13—C12	1.378 (4)
C11—C16	1.381 (4)	C13—H13A	0.9300
C11—C12	1.383 (4)	C12—H12A	0.9300
C11—C10	1.503 (4)	C6—H6A	0.9300
C10—H10A	0.9700	C2—C3	1.349 (5)
C10—H10B	0.9700	C2—C1	1.405 (4)
C9—C7	1.369 (4)	C2—H2A	0.9300

C4—C5	1.409 (4)	C3—H3A	0.9300
C4—C3	1.410 (4)	C1—H1A	0.9300
C16—C15	1.390 (4)	C23—N2	1.139 (5)
C16—H16A	0.9300	C21—C20	1.365 (5)
C17—C18	1.378 (4)	C21—H21A	0.9300
C17—C22	1.405 (4)	C18—C19	1.385 (5)
C17—C14	1.494 (4)	C18—H18A	0.9300
C14—C15	1.383 (4)	C20—C19	1.382 (6)
C14—C13	1.393 (4)	C20—H20A	0.9300
C7—C6	1.410 (4)	C19—H19A	0.9300
C7—H7A	0.9300		
C9—O1—C10	116.0 (2)	C21—C22—C23	119.4 (3)
C1—N1—C8	116.8 (2)	C17—C22—C23	120.0 (3)
N1—C8—C4	123.0 (2)	C14—C15—C16	120.8 (3)
N1—C8—C9	118.6 (2)	C14—C15—H15A	119.6
C4—C8—C9	118.4 (2)	C16—C15—H15A	119.6
C16—C11—C12	118.5 (3)	C12—C13—C14	120.8 (3)
C16—C11—C10	124.0 (2)	C12—C13—H13A	119.6
C12—C11—C10	117.4 (2)	C14—C13—H13A	119.6
O1—C10—C11	110.8 (2)	C13—C12—C11	120.9 (3)
O1—C10—H10A	109.5	C13—C12—H12A	119.5
C11—C10—H10A	109.5	C11—C12—H12A	119.5
O1—C10—H10B	109.5	C5—C6—C7	120.6 (3)
C11—C10—H10B	109.5	C5—C6—H6A	119.7
H10A—C10—H10B	108.1	C7—C6—H6A	119.7
O1—C9—C7	124.9 (3)	C3—C2—C1	118.8 (3)
O1—C9—C8	115.3 (2)	C3—C2—H2A	120.6
C7—C9—C8	119.8 (2)	C1—C2—H2A	120.6
C5—C4—C3	123.1 (3)	C2—C3—C4	120.0 (3)
C5—C4—C8	120.1 (3)	C2—C3—H3A	120.0
C3—C4—C8	116.8 (3)	C4—C3—H3A	120.0
C11—C16—C15	120.7 (3)	N1—C1—C2	124.5 (3)
C11—C16—H16A	119.6	N1—C1—H1A	117.7
C15—C16—H16A	119.6	C2—C1—H1A	117.7
C18—C17—C22	118.0 (3)	N2—C23—C22	178.8 (4)
C18—C17—C14	120.7 (3)	C20—C21—C22	120.0 (3)
C22—C17—C14	121.3 (3)	C20—C21—H21A	120.0
C15—C14—C13	118.2 (3)	C22—C21—H21A	120.0
C15—C14—C17	122.2 (3)	C17—C18—C19	121.2 (4)
C13—C14—C17	119.6 (2)	C17—C18—H18A	119.4
C9—C7—C6	120.9 (3)	C19—C18—H18A	119.4
C9—C7—H7A	119.5	C21—C20—C19	120.0 (3)
C6—C7—H7A	119.5	C21—C20—H20A	120.0
C6—C5—C4	120.2 (3)	C19—C20—H20A	120.0
C6—C5—H5A	119.9	C20—C19—C18	120.2 (4)
C4—C5—H5A	119.9	C20—C19—H19A	119.9
C21—C22—C17	120.6 (3)	C18—C19—H19A	119.9
C1—N1—C8—C4	1.3 (4)	C18—C17—C22—C23	178.7 (3)

C1—N1—C8—C9	−178.8 (3)	C14—C17—C22—C23	1.2 (4)
C9—O1—C10—C11	−171.6 (2)	C13—C14—C15—C16	1.7 (5)
C16—C11—C10—O1	8.7 (4)	C17—C14—C15—C16	−178.0 (3)
C12—C11—C10—O1	−174.3 (2)	C11—C16—C15—C14	−0.4 (5)
C10—O1—C9—C7	−0.4 (4)	C15—C14—C13—C12	−1.2 (5)
C10—O1—C9—C8	−179.8 (2)	C17—C14—C13—C12	178.6 (3)
N1—C8—C9—O1	3.3 (3)	C14—C13—C12—C11	−0.7 (5)
C4—C8—C9—O1	−176.8 (2)	C16—C11—C12—C13	2.0 (4)
N1—C8—C9—C7	−176.2 (3)	C10—C11—C12—C13	−175.2 (3)
C4—C8—C9—C7	3.7 (4)	C4—C5—C6—C7	0.8 (5)
N1—C8—C4—C5	177.1 (3)	C9—C7—C6—C5	0.2 (5)
C9—C8—C4—C5	−2.7 (4)	C1—C2—C3—C4	0.4 (5)
N1—C8—C4—C3	−2.5 (4)	C5—C4—C3—C2	−178.1 (3)
C9—C8—C4—C3	177.6 (2)	C8—C4—C3—C2	1.5 (4)
C12—C11—C16—C15	−1.4 (5)	C8—N1—C1—C2	0.9 (5)
C10—C11—C16—C15	175.6 (3)	C3—C2—C1—N1	−1.8 (5)
C18—C17—C14—C15	111.6 (4)	C21—C22—C23—N2	54 (24)
C22—C17—C14—C15	−71.0 (4)	C17—C22—C23—N2	−125 (23)
C18—C17—C14—C13	−68.1 (4)	C17—C22—C21—C20	−0.5 (5)
C22—C17—C14—C13	109.3 (3)	C23—C22—C21—C20	−179.0 (3)
O1—C9—C7—C6	178.1 (3)	C22—C17—C18—C19	0.7 (5)
C8—C9—C7—C6	−2.5 (4)	C14—C17—C18—C19	178.3 (3)
C3—C4—C5—C6	−179.8 (3)	C22—C21—C20—C19	0.0 (5)
C8—C4—C5—C6	0.5 (4)	C21—C20—C19—C18	0.9 (6)
C18—C17—C22—C21	0.1 (4)	C17—C18—C19—C20	−1.3 (6)
C14—C17—C22—C21	−177.4 (3)		

Fig. 1

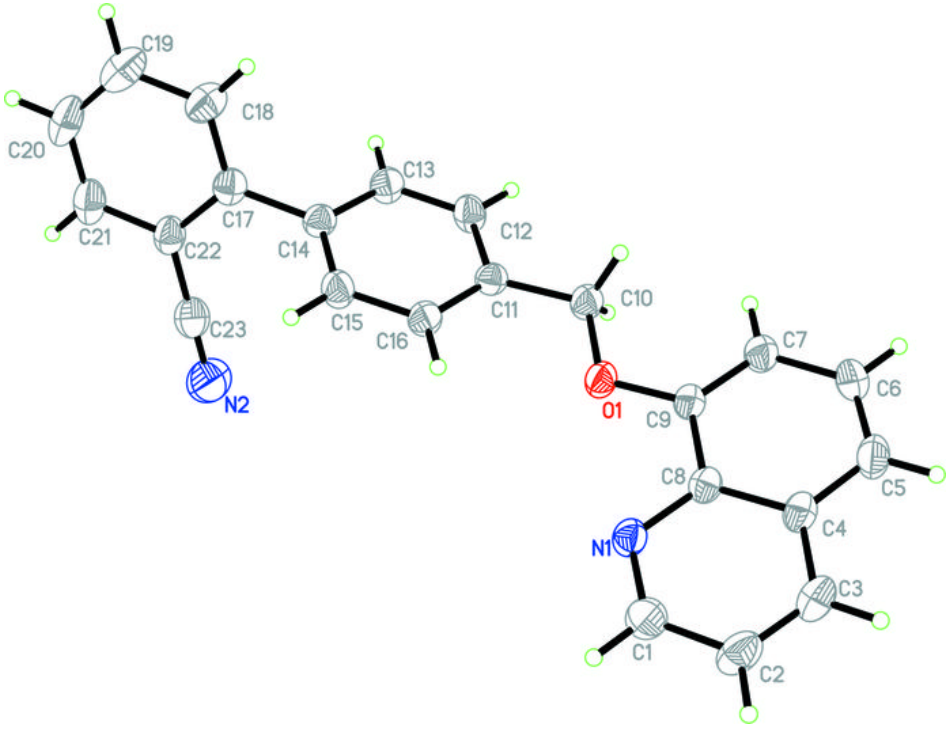

C19
C18
C20
C13
C17
C14
C12
C21
C22
C10
C11
C23
C15
C16
C7
C6
N2
O1
C9
C8
C5
C4
N1
C3
C1
C2

Fig. 2